

## SYNTHESIS AND ANTIMICROBIAL EVALUATION OF 3,5-PYRAZOLIDINE-DIONE SUBSTITUTED 4-QUINOLONE DERIVATIVES

MANISH GUPTA<sup>A</sup>, NEERAJ UPMANYU\*<sup>A</sup>, SOMA PRAMANIK<sup>A</sup>, CHANDRA KISHORE TYAGI<sup>b</sup> AND AMOL CHANDEKAR<sup>A</sup>

<sup>a</sup>Department of Pharmaceutical Chemistry, R.K.D.F. College of Pharmacy, Bhopal, (M.P.)-462047

<sup>b</sup>Department of Pharmacology, RKDF College Of Pharmacy Bhopal (M.P.)-462047

### Abstract

A series of novel 1-(2-methyl-4-oxo-1,4-dihydroquinoline-6-carbonyl)-2-(substituted phenyl)-pyrazolidine-3, 5-diones [5A-5H] were synthesized with different aromatic hydrazides and evaluated for antibacterial and antifungal activity. The structures of the compounds were confirmed by nitrogen analysis, FT-IR, <sup>1</sup>H NMR and mass spectral data. The antimicrobial activity of synthesized compounds were examined against two gram positive bacteria (*S. aureus*, *B. subtilis*), two gram negative bacteria (*E. coli*, *S. species*) and two fungi (*C. albicans* and *A. niger*) using broth dilution technique. Some of the synthesized compounds exhibited mild to moderate antibacterial and antifungal activity.

### Key words:

Antimicrobial, 4-quinolone, 3, 5-pyrazolidine-dione

### How to Cite this Paper:

Manish Gupta, Neeraj Upmanyu\*, Soma Pramanik and Amol Chandekar "Synthesis and Antimicrobial evaluation of 3,5-pyrazolidine-Dione substituted 4-Quinolone derivatives", Int J. Drug Dev. & Res., April-June 2011, 3(2): 233-239

\*Corresponding author, Mailing address:  
Neeraj Upmanyu  
Email- [neerajupmanyu2007@gmail.com](mailto:neerajupmanyu2007@gmail.com)

**Copyright © 2010 IJDDR, Neeraj Upmanyu et al.** This is an open access paper distributed under the copyright agreement with Serials Publication, which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

**Article History:**-----

**Date of Submission: 18-04-2011**

**Date of Acceptance: 22-05-2011**

**Conflict of Interest: NIL**

**Source of Support: NONE**

### Introduction

The rapid rise in microbial resistance to the traditional antibiotics has necessitated a continuing search for new classes of compounds with novel modes of antimicrobial activity<sup>[1]</sup>. The 4-quinolone antimicrobials have a number of advantages over other classes of antimicrobial agents. They have wide spectrum of activity, well absorbed orally, have relatively long serum half lives and minimal toxicity<sup>[2]</sup>. The currently used quinolone derivatives are known to have several drug interactions and adverse side effects<sup>[3]</sup>. Recently 3, 5- dioxo pyrazolidines were shown to be novel inhibitors of UDP-N-Acetylenolpyruvylglucosamine reductase (MurB) with activity against Gram +Ve bacteria<sup>[4]</sup>.

Similarly a variety of biological activity of substituted 4-Quinolone have been reported and these includes antitubercular<sup>[5]</sup>, antitumor<sup>[6]</sup>, antifilarial<sup>[7]</sup>, anti-HIV<sup>[8]</sup> and antimicrobial activity of chloroquinoline containing pyrazoline<sup>[9]</sup>. Keeping these above facts in view, we considered it for interest to synthesize a few 3,5-pyrazolidinedione substituted 4-quinolone for their antimicrobial activity.

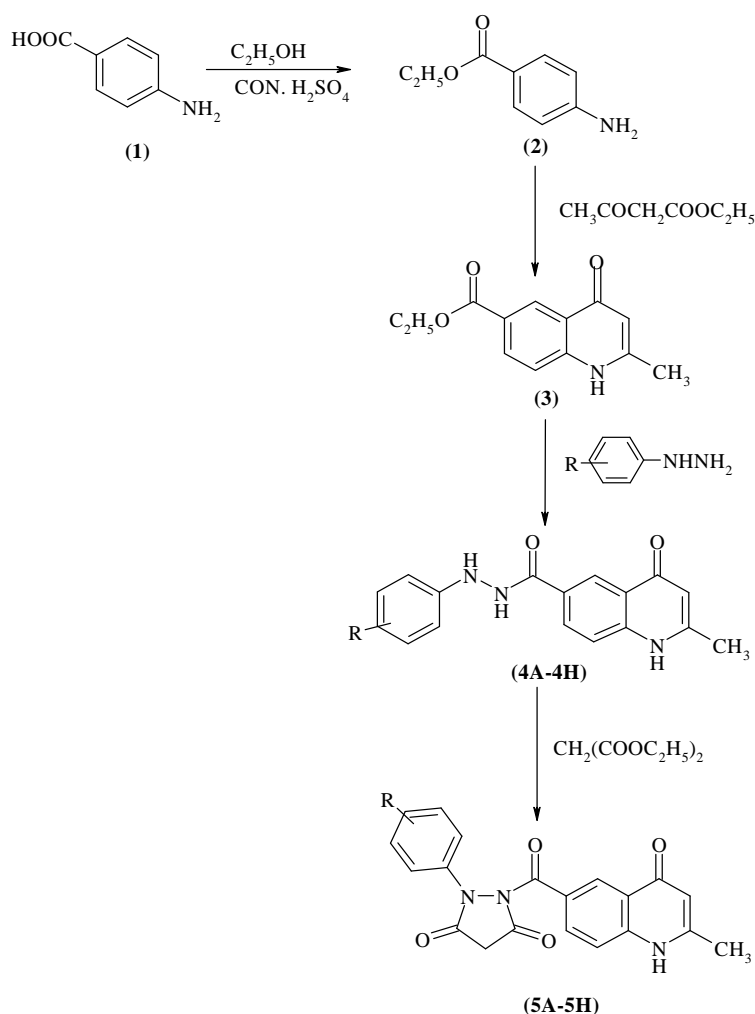
## Experimental

### Chemistry

The chemicals and reagents used in the present project were of AR and LR grade, procured from Aldrich, Hi-Media, Merck, Sigma and Ranbaxy. Melting point of the synthesized compounds were

determined by open capillary method and is uncorrected. The purity of compounds was checked by TLC on microplates using silica-gel-G, solvent system was Toluene: Ethanol (95:5) with iodine vapors as detecting agent. Nitrogen estimation was done using elemental analyzer Heraeus carlo Erba 1108, IR spectra of the synthesized compounds were recorded on Shimadzu 8400S FT-IR spectrometer (KBr disc). <sup>1</sup>H NMR spectra were recorded on amx-400 NMR spectrometer (D<sub>2</sub>O, TMS) and Mass spectra recorded on Jeol Sx 102/DA-600 mass spectrometer/Data System using fast moving bombardment (FAB) technique.

The title compounds [5A-H] were prepared through several steps as depicted in Fig 1.



**Figure 1:** Synthesis of the title compounds (5A-H)

### Ethyl p-aminobenzoate [2]

Ethyl p-aminobenzoate was prepared by Esterification of PABA by using the reported method<sup>[10]</sup>. *P*-amino benzoic acid 0.04 mol (5.48 g) was dissolved in 75mL of ethanol and mixture was heated on a sand bath until the entire solid dissolved. Cooled in room temperature and conc. Sulfuric acid (12.5 mL) was added. A large amount of precipitate formed when sulfuric acid was added, but dissolved during reflux, an air condenser was attached and refluxed for 60-70 min. than mixture was allowed to cool at room temperature and a solution of sodium bicarbonate (10 %) was added to neutralize the excess sulfuric acid. As pH increased, a white precipitate of ethyl *p*-amino benzoate (benzocaine) was produced. When gas no longer evolved as drop of sodium bicarbonate (pH 8). Prepared product was collected by vacuum filtration. Product was dried in open container and purified from ethanol. Yield: 94.4 %, m.p. 99-101°C.

### 2-methyl-4-oxo-1,4 dihydro-quinoline-6-carboxylic acid ethyl ester [3]

The compound 2-methyl-4-oxo-1,4 dihydro quinoline-6-carboxylic acid ethyl ester was prepared from Ethyl p-amino benzoate by Conrad-Limpach reaction<sup>[11]</sup>. A mixture of ethyl p-amino benzoate [2] 0.07 mol (11.55 g), ethyl acetoacetate 0.1 mol (13 g), dioxane (25 mL) and trace of conc. HCl was taken in 250 mL round bottom flask and refluxed at 70-80°C for 3 h, cooled at room temperature and conc. sulfuric acid (20 mL) was added. Than the mixture was refluxed at 190-200°C for 1 h and hot mixture was poured in 500mL of ice/cold water with constant stirring. Separated solid was filtered, dried, and recrystallized from ethanol. Yield: 76.92%, m.p. 265-267°C.

### 2-methyl-4-oxo-1,4 dihydro quinoline-6carboxylic acid N-phenylhydrazides [4A-H]:

A suspension of 2-methyl-4-oxo-1,4 dihydro-quinoline-6-carboxylic acid ethyl ester [3] 0.026mol (6 g) in methanol (10 mL) was prepared. Required

phenyl hydrazine 0.05 mol (6 mL) then added to it at room temperature. After stirring, ethanol (20 mL) was added and refluxed for 1 h, cooled at room temperature, filtered and washed the solid with diethyl ether (20 mL) and purified from ethanol<sup>[12]</sup>. Some common stretching and bending vibrations are as follows IR (KBr) 3282 (N-H), 3039 (Ar, C-H), 2957 (alkanes), 1720 (C=O), 1631,1548,1242 (CONH) 1606 (>C=O, quinolone). The m.p. and % yield of the synthesized compounds were given in Table 1.

### 1-(2-methyl-4-oxo-1, 4-dihydro quinoline-6-carbonyl)-2-phenyl-pyrazolidine-3, 5-dione [5A-H]:

A mixture of 2-methyl-4-oxo-1,4 dihydro quinoline-6carboxylic acid-N'-phenyl hydrazides (0.01mol) [4A-H] were prepared. Diethyl malonate (0.015 mol), ethanol (90mL) and acetic acid (1mL) was added and refluxed for 5hrs. The reaction mixture was left in open dish for 2-3 hrs. The solid precipitate formed was filtered, dried and recrystallized from ethanol<sup>[13]</sup>. The m.p. and % yield of the synthesized compounds were given in Table 1, Physical data in Table 2 and spectral data of the titled synthesized compounds [5A-H] were given in Table 3.

**Table 1:** List of the synthesized compounds

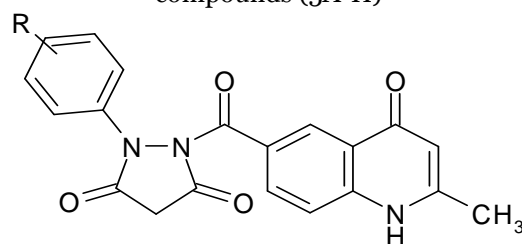
Compound code	R	m.p (°C)	% Yield
4A	H	265-267	76.92
4B	2,4,6-Br	134-136	51
4C	2,6-Cl	220-222	61.5
4D	4-Cl	125-127	59.32
4E	2,4-NO <sub>2</sub>	175-177	57.80
4F	2-NO <sub>2</sub>	156-158	61.47
4G	4-NO <sub>2</sub>	205-207	57.30
4H	4-CH <sub>3</sub>	118-120	64.50
5A	H	230-232	55.55
5B	2,4,6-Br	205-207	66.66
5C	2,6-Cl	285-287	57.50
5D	4-Cl	200-202	41.66
5E	2,4-NO <sub>2</sub>	210-212	57.14
5F	2-NO <sub>2</sub>	205-207	55.55
5G	4-NO <sub>2</sub>	205-207	55.55
5H	4-CH <sub>3</sub>	185-187	50

**Table- 2:** Physical data of Titled synthesized compounds (5A-H)

Compound code	R	Mol. Formula	Mol. Wt. (g)	Nitrogen estimation found (calculated)	<sup>a</sup> R <sub>f</sub>
5A	H	C <sub>20</sub> H <sub>15</sub> O <sub>4</sub> N <sub>3</sub>	361.52	11.32 (11.63)	0.32
5B	2,4,6-Br	C <sub>20</sub> H <sub>12</sub> O <sub>4</sub> N <sub>3</sub> Br <sub>3</sub>	597.48	6.94 (7.03)	0.39
5C	2,6-Cl	C <sub>20</sub> H <sub>13</sub> O <sub>4</sub> N <sub>3</sub> Cl <sub>3</sub>	430.03	9.69 (9.73)	0.17
5D	4-Cl	C <sub>20</sub> H <sub>14</sub> O <sub>2</sub> N <sub>3</sub> Cl	395.58	10.51(10.60)	0.32
5E	2,4-NO <sub>2</sub>	C <sub>20</sub> H <sub>13</sub> O <sub>8</sub> N <sub>5</sub>	451.11	15.63 (15.52)	0.47
5F	2-NO <sub>2</sub>	C <sub>20</sub> H <sub>14</sub> O <sub>6</sub> N <sub>4</sub>	406.35	13.84 (13.25)	0.40
5G	4-NO <sub>2</sub>	C <sub>20</sub> H <sub>14</sub> O <sub>6</sub> N <sub>4</sub>	406.35	13.24 (13.79)	0.52
5H	4-CH <sub>3</sub>	C <sub>21</sub> H <sub>17</sub> O <sub>2</sub> N <sub>3</sub>	375.39	11.11 (11.19)	0.35

<sup>a</sup>Solvent system used was : Toluene: Ethanol (95:5,v/v)

**Table 3:** Spectral data of titled synthesized compounds (5A-H)



Comp. ce	IR (cm <sup>-1</sup> ) KBr	<sup>1</sup> H NMR δ (D <sub>2</sub> O, TMS)	FAB Mass m/z
5A	3062 (Ar, C-H), 2943 (alkanes), 1720 (C=O), 1614 (>C=O, quinolone), 1690 (>C=O, pyrazolidinedione)	2.36 (s, 3H, CH <sub>3</sub> ), 3.39 (s, 2H, CH <sub>2</sub> ) 6.21 (s, 1H, Quin), 7.05-7.20 (m, 2H, Ar-H), 7.26 (s, 1H, Ar-H), 7.54-7.69 (m, 3H, Ar-H), 8.06 (s, 1H, Ar-H), 8.45 (s, 1H, Ar-H), 9.21 (s, 1H, NH, Quin)	361 (M+H) <sup>+</sup>
5B	3060 (Ar, C-H), 2983 (alkanes), 1720 (C=O), 1614 (>C=O, quinolone), 1693 (>C=O, pyrazolidinedione), 1093 (C-Br)	2.23 (s, 3H, CH <sub>3</sub> ), 3.58 (s, 2H, CH <sub>2</sub> ) 6.21 (s, 1H, Quin), 7.17-7.19 (d, J=8 Hz, 1H, Ar-H), 7.85 (s, 2H, Ar-H), 8.23-8.25 (d, J=8 Hz, 1H, Ar-H), 8.73 (s, 1H, Ar-H), 9.28 (s, 1H, NH, Quin)	597 (M+H) <sup>+</sup>
5C	3060 (Ar, C-H), 2983 (alkanes), 1720 (C=O), 1614 (>C=O, quinolone), 1683 (>C=O, pyrazolidinedione), 757 (C-Cl)	2.25 (s, 3H, CH <sub>3</sub> ), 3.41 (s, 2H, CH <sub>2</sub> ) 6.33 (s, 1H, Quin), 7.18-7.24 (t, 1H, Ar-H) 7.28-7.30 (d, J=8 Hz, 1H, Ar-H), 7.66-7.68 (d, J=8 Hz, 2H, Ar-H), 8.21-8.23 (d, J=8 Hz, 1H, Ar-H), 8.69 (s, 1H, Ar-H), 9.25 (s, 1H, NH, Quin)	429 (M+H) <sup>+</sup>
5D	3045 (Ar, C-H), 2983 (alkanes), 1720 (C=O), 1614 (>C=O, quinolone), 1685 (>C=O, pyrazolidinedione), 757 (C-Cl)	2.29 (s, 3H, CH <sub>3</sub> ), 3.49 (s, 2H, CH <sub>2</sub> ) 6.28 (s, 1H, Quin), 7.21-7.23 (d, J=8 Hz, 2H, Ar-H), 7.24-7.26 (d, J=8 Hz, 1H, Ar-H), 7.77-7.79 (d, J=8 Hz, 2H, Ar-H), 8.19-8.21 (d, J=8 Hz, 1H, Ar-H), 8.48 (s, 1H, Ar-H), 9.33 (s, 1H, NH, Quin)	395 (M+H) <sup>+</sup>
5E	3060 (Ar, C-H), 2983 (alkanes), 1720 (C=O), 1614 (>C=O, quinolone), 1685 (>C=O, pyrazolidinedione), 1514 (N=O)	2.35 (s, 3H, CH <sub>3</sub> ), 3.38 (s, 2H, CH <sub>2</sub> ) 6.14 (s, 1H, Quin), 7.11-7.193 (d, J=8 Hz, 1H, Ar-H) 8.35-8.37 (d, J=8 Hz, 2H, Ar-H), 8.71-8.79 (m, 2H, Ar-H), 8.88 (s, 1H, Ar-H), 9.51 (s, 1H, NH, Quin)	451 (M+H) <sup>+</sup>
5F	3058 (Ar, C-H), 2983 (alkanes), 1720 (C=O), 1612 (>C=O, quinolone), 1685 (>C=O, pyrazolidinedione), 1514 (N=O)	2.21 (s, 3H, CH <sub>3</sub> ), 3.32 (s, 2H, CH <sub>2</sub> ) 6.17 (s, 1H, Quin), 7.19-7.21 (d, J=8 Hz, 1H, Ar-H), 7.36-7.45 (m, 2H, Ar-H), 7.89-7.91 (d, J=8 Hz, 1H, Ar-H), 8.18-8.25 (m, 2H, Ar-H), 8.41 (s, 1H, Ar-H), 9.53 (s, 1H, NH, Quin)	406 (M+H) <sup>+</sup>
5H	3043 (Ar, C-H), 2945, 2981 (alkanes), 1720 (C=O), 1612 (>C=O, quinolone), 1687 (>C=O, pyrazolidinedione)	2.30 (s, 3H, CH <sub>3</sub> ), 3.46 (s, 2H, CH <sub>2</sub> ) 6.23 (s, 1H, Quin), 7.05-7.07 (d, J=8 Hz, 2H, Ar-H), 7.26-7.28 (d, J=8 Hz, 1H, Ar-H), 7.83-7.85 (d, J=8 Hz, 2H, Ar-H), 8.11-8.13 (d, J=8 Hz, 1H, Ar-H), 8.65 (s, 1H, Ar-H), 9.48 (s, 1H, NH, Quin)	375 (M+H) <sup>+</sup>

### Antimicrobial evaluation

The synthesized compounds were evaluated against bacterial strains i.e. *Bacillus subtilis*, *Staphylococcus aureus*, *Escherichia coli* and *Shigella* species and fungal strains i.e. *Candida albicans* and *Aspergillus niger* using broth dilution method. The strains were procured from MTCC, Chandigarh, as clinical isolates. Ciprofloxacin and Fluconazole were used as standards for antibacterial and antifungal activity studies respectively.

### Antibacterial Testing by Broth Dilution Method

The stock solution was prepared in dimethylsulphoxide (DMSO). The solutions in the test medium furnished the required concentration ranging from 20 to 160 µg/mL. Nutrient broth<sup>[14]</sup> (beef extract 3 g, peptone 5 g, sodium chloride 5 g and distilled water-q.s. to 1000 mL) was employed as culture media for antibacterial studies. The

sterilization of the nutrient broth, culture tubes, pipette and other glassware was done by autoclaving at 15 lb/sq inch pressure for 30 min. Incubation was carried out at  $37\pm 1^\circ\text{C}$  for 48 h. The MIC value were recorded as the lowest concentration of the substances that had no visible turbidity using a systronics Turbidometer. All the tests were performed in triplicate for determination of MICs.

#### Antifungal Testing by Broth Dilution Method

Similarly antifungal activity was carried out. For antifungal studies potato dextrose broth [15] (potato extract 250g, dextrose 20g and distilled water q.s. 1000 mL) was employed. Incubation was carried out at  $28\pm 1^\circ\text{C}$  for 72 h. All the tests were performed in triplicate for determination of MICs.

#### Results and discussion

A series of novel 1-(2-methyl-4-oxo-1,4-dihydro quinoline-6-carbonyl)-2-(substituted Phenyl)-pyrazolidine-3, 5-diones [5A-5H] were synthesized with different aromatic hydrazides. Para amino benzoic acid (PABA) [1] was converted into Ethyl p-amino benzoate [2] by its esterification. The melting point of PABA is  $188-190^\circ\text{C}$ . The melting point of the ethyl p-amino benzoate is  $99-101^\circ\text{C}$ . The difference in melting point clearly indicates the formation. IR spectrum of the compound [2] clearly shows the  $\nu_{\text{max}}$  at  $1311\text{ cm}^{-1}$  and  $1720\text{ cm}^{-1}$  due to ester (C-O) and ketone (C=O) respectively. Ethyl p-amino benzoate [2] is converted into 2-methyl-4-oxo-1,4-dihydro quinoline-6-carboxylic acid ethyl ester [3] by Conrad-Limpach reaction. The IR spectrum shows carbonyl group ( $>\text{C}=\text{O}$  of 4-quinolone) at  $1606\text{ cm}^{-1}$  and  $2^\circ$  amine (-NH-) at  $3284\text{ cm}^{-1}$  which clearly indicate formation of the compound [3].

Compound 2-methyl-4-oxo-1,4-dihydro-quinoline-6-carboxylic acid-N-Phenyl hydrazide [4A-H] is prepared from [3] with different substituted phenyl hydrazine. Some important peaks of IR spectrum of [4A-H] shows  $\nu_{\text{max}}$  at  $1631\text{ cm}^{-1}$ ,

$1548\text{ cm}^{-1}$  and  $1242\text{ cm}^{-1}$  due to amide (-CONH-) and  $1614\text{ cm}^{-1}$  due to ( $>\text{C}=\text{O}$ , quinolone),  $1531\text{ cm}^{-1}$  (N=O asymmetric str),  $1332\text{ cm}^{-1}$  (N=O symmetric str) due to  $-\text{NO}_2$  group,  $757\text{ cm}^{-1}$  due to (C-Cl) and  $2940\text{ cm}^{-1}$  due to (C-H str of methyl group on benzene ring). In the last step, title compounds [5A-H] was prepared from [4A-H] with diethyl malonate. The IR spectrum of [5A-H] clearly indicates carbonyl group ( $>\text{C}=\text{O}$  of pyrazolidine-3, 5-dione) at  $1690\text{ cm}^{-1}$ .

In  $^1\text{H-NMR}$  spectra, presence of NMR signals at  $\delta$  2.1 ppm (3H,  $\text{CH}_3$ ), 3.4 ppm (2H, pyrazolidinedione), 4.67 ppm (1H, NH), 6.86 ppm (1H  $\text{C}_3\text{-H}$ ), 7.22-7.26 ppm (8H, Ar) clearly indicates the formation.

The synthesized compounds were subjected to antimicrobial studies. MICs were determined by the broth dilution method.

#### Antibacterial activity

The antibacterial screening was done using *B. subtilis*, *S. aureus*, *E. coli* and *S. species*. Ciprofloxacin was selected as standard drug. Out of all the eight compounds evaluated for antibacterial studies, compound 5G showed appreciable antibacterial activity against all four bacterial strains, ( $66\mu\text{g/mL}$  against *S. aureus*,  $64\mu\text{g/mL}$  against *B. subtilis*,  $68\mu\text{g/mL}$  against *S. species* and  $66\mu\text{g/mL}$  against *E. coli*). Other compound with some moderate activity was 5A ( $88\mu\text{g/mL}$  against *S. aureus*,  $78\mu\text{g/mL}$  against *B. subtilis*,  $82\mu\text{g/mL}$  against *S. species* and  $98\mu\text{g/mL}$  against *E. coli*) and 5E ( $86\mu\text{g/mL}$  against *S. aureus*,  $74\mu\text{g/mL}$  against *B. subtilis*,  $90\mu\text{g/mL}$  against *S. species* and  $88\mu\text{g/mL}$  against *E. coli*). MIC of the tested synthesized compounds are shown in Table 4.

**Table 4:** Antibacterial activity of Synthesized compounds

Compound	MIC (µg/mL)			
	<i>S.aureus</i> (Gram +ve)	<i>B.subtilis</i> (Gram +ve)	<i>E.coli</i> (Gram -ve)	<i>S. species</i> (Gram -ve)
5A	88	78	98	82
5B	106	112	106	120
5C	102	102	104	98
5D	94	102	92	120
5E	86	74	88	90
5F	118	88	114	96
5G	66	64	66	68
5H	102	118	118	106
Ciprofloxacin	8	10	12	6

#### Antifungal activity

Similarly antifungal activity was carried out using fungal strains *A. niger* and *C. albicans*. Fluconazole was selected as the standard drug. Out So far as antifungal activity is concerned, the most effective compounds were **5F** (62 µg/mL against *C. albicans*, 66 µg/mL against *A. niger* , **5G** (78 µg/mL against *C. albicans*, 64 µg/mL against *A. niger*.) All compounds have shown moderate antifungal activity against *C. albicans* and *A. niger*. Other compounds with moderate antifungal activity were **5A** (78 µg/mL), **5H** (82 µg/mL) and **5C** (88 µg/mL). MIC of the tested synthesized compounds are shown in Table 5.

**Table 5 :** Antifungal activity of Synthesized compounds

Compound	MIC (µg/mL)	
	<i>C. albicans</i>	<i>A. niger</i>
5A	78	102
5B	104	96
5C	88	92
5D	108	112
5E	112	104
5F	62	66
5G	78	64
5H	82	102
Fluconazole	6	10

SAR studies of these synthesized compounds also revealed that substitution on benzene ring system which was attached to pyrazolidinedione also affect the activity. Compound with nitro substitution at benzene ring provides more effective compound than alkyl and halogen substitution. Mono substitution is more effective than 'di' or 'tri' substitution.

#### Conclusion

It can be concluded that synthesized fused heterocyclic ring display not only good antifungal activity but also provide good antibacterial activity as well. The effectiveness of the synthesized compounds as antibacterial was found to be greater, when compared to antifungal activity. Compound with nitro substitution on benzene ring is more preferable. These compounds can be considered as lead molecules for future investigations.

#### Acknowledgement

Authors are thankful to the Heads, Department of Pharmaceutical Chemistry and Department of Microbiology for kind permission to carry out the work in their departments. The help rendered by IISC, Quest Bangalore for spectral and nitrogen analysis is gratefully acknowledged.

#### References

- 1) Goodman & Gilman's The Pharmacological Basis of Therapeutics, 10th edn, (Hardman JG, Limbird LE, Gilman AG, eds), Merck & Co., Inc., Whitehouse station, N.J., 2001, pp 1065-1066.
- 2) Bansal RK. Heterocyclic Chemistry, 3rd edn, New Age Publishers, New Delhi, 1999, pp 477-478.
- 3) Block JH, Beale JM. Jr. Willson and Gisvold's Textbook of Organic and Pharmaceutical Chemistry, 11th edn, A Wolters Kluwer Company; Lippincott Williams & Wilkins, New York, 2004, pp 217-218.
- 4) Yang Y, Severin A, Chopra R, Krishnamurthy G, Singh G, Hu W, Keeney D, Svenson K, Petersen PJ, Labthavikul P, Shlaes DM, Rasmussen BA, Failli AA, Shumsky JS, Kutterer KMK, Gilbert A and Mansour TS. 3,5-Dioxypyrazolidines, novel inhibitors of UDP-N-acetylenolpyruvylglucosamine reductase (MurB) with activity against gram-positive bacteria. Antimicrobial Agents and Chemotherapy 2006; 50(2): 556-564.
- 5) Shindikar AV, Viswanathan CL. Novel fluoroquinolones: Design, synthesis, and *in vivo*

- activity in mice against mycobacterium tuberculosis H<sub>37</sub>Rv. *Bioorganic and Medicinal Chemistry* 2005; 15: 1803-1806.
- 6) Tabarrini O, Cecchetti V, Fravolini A, Nocentini G, Barzi A, Sabatin S, Miao H and Siss C. Design and synthesis of modified quinolones as antitumoral acridones. *Journal of Medicinal Chemistry* 1999; 42(12): 2136.
- 7) Shrivastava SK, Chuhan MS, Bhaduri AP, Fatima N, Chatterjee RK. Quinolones: novel probes in antifilarial chemotherapy. *Journal of Medicinal Chemistry* 2000; 43: 2275-2279.
- 8) Cecchetti V, Parolin C, Moro S, Pecere T, Filippini V, Calistri A, Tabarrini O, Gatto B, Palumbo M, Fravolini A and Palù G. 6-Amino quinolones as new potential anti-HIV agents. *Journal of Medicinal Chemistry* 2000; 43: 3799-3802.
- 9) Bawa S, Kumar S, Drabu S, Panda BP, Kumar R. Synthesis and antimicrobial activity of 2-chloroquinoline incorporated pyrazoline derivatives. *Journal of Pharmacy and Bioallied Sciences* 2009; 01: 32-36.
- 10) Furniss BS, Hannaford AJ, Smith PWG, Tatchell AR. *Vogel's Textbook of Practical Organic Chemistry*, 5th edn, Longman Scientific and Technical, London, 1989, pp 1205-1206.
- 11) Katritzky AR, Pozharski AP. *Hand Book of Heterocyclic Chemistry*, 2nd edn, Pergamone/Elsevier, 2000, pp 4-5.
- 12) Deshmukh MB, Jagtap SS and Deshmukh SA. Solvent free accelerated synthesis of 2-hydrazinobenzothiazole derivatives using microwave. *Journal of Indian Chemical Society* 2006; 83: 1055-1057.
- 13) Al-Soud YA and Al-Masoudi NA. A new class of dihaloquinolones bearing *N'*-aldehydoglycosylhydrazides, mercapto-1,2,4-triazole, oxadiazoline and  $\alpha$ -amino ester precursors: synthesis and antimicrobial activity. *Journal of the Brazilian Chemical Society* 2003; 14(5): 790-796.
- 14) Aneja KR. *Experiments in Microbiology and Plant Pathology, Tissue Culture and Mushroom Production Technology*, 3rd edn, New Age International Publishers, New Delhi, 2002, pp 568-570.
- 15) Johnston A, Booth C. *Plant Pathologist's Pocketbook*, 2nd edn, Oxford & IBH Publishing Co., New Delhi, 1983, p. 439.

